

PHOSPHOLIPASE A₂ INHIBITION BY MANOALIDE:
DEVELOPMENT OF SIMPLE ANALOGUES AND
NECESSARY FUNCTIONAL GROUPS FOR INHIBITION*)

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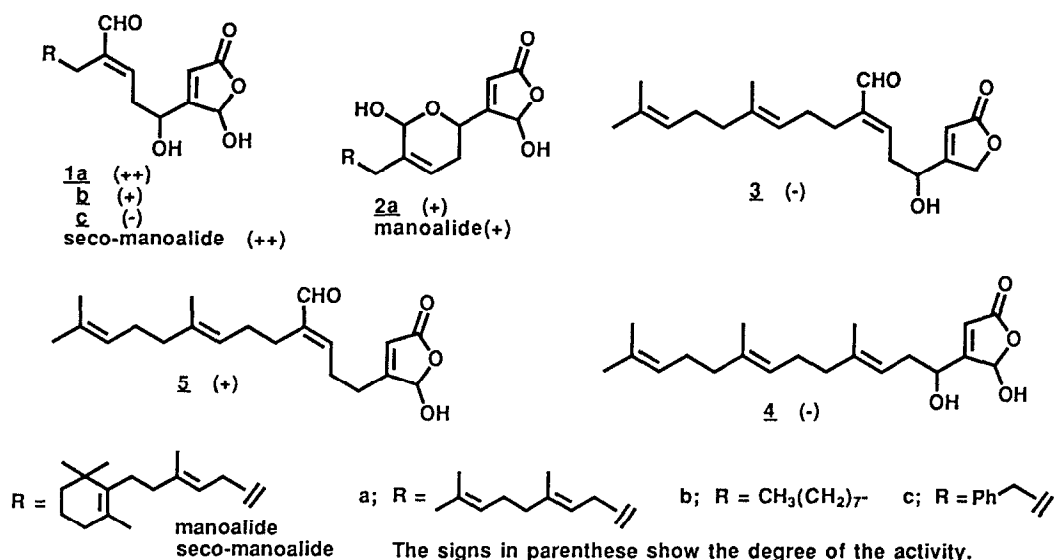
Abstract: Manoalide analogue 1a inhibited bovine pancreatic phospholipase A₂ to the same extent as seco-manoalide. Two aldehyde groups in a manoalide molecule are indispensable to the enzyme inhibition.

Manoalide, which was isolated from a marine sponge *Luffariella variabilis*,¹ showed anti-inflammatory properties *in vivo*.² Phospholipase A₂(PLA₂), which hydrolyzes the ester linkage at the C-2 position of glycerophospholipid, is known as one of the key enzymes in the metabolism of arachidonic acid.³ Two kinds of PLA₂s, isolated from cobra⁴ and bee⁵ venoms, were shown to be inhibited by natural manoalide, and their inhibition mechanisms were proposed. Recently, we found independently similar inactivation phenomena of bovine pancreatic PLA₂ by synthetic manoalide and its analogues, and started to elucidate the inhibition mechanism. Bovine pancreatic PLA₂ is thought to be one of the most suitable materials for such purposes, since a great deal of information concerning the three dimensional structure⁶ and kinetic properties of the enzyme are available.

Our studies on the enzymatic inhibition of bovine pancreatic PLA₂ by manoalide have been focused upon the following four points. 1. Development of a simple analogue for manoalide which possesses an equal inhibitory ability to that for *d,l*-manoalide. 2. Determination of the functional groups in the manoalide molecule which are responsible for the enzyme inhibition. 3. Identification and characterization of the amino acid residues of this PLA₂ which are modified by a manoalide analogue. 4. Understanding of the reaction mechanism of manoalide with these amino acid residues. In this paper, we report the development of a manoalide analogue which is as effective as *d,l*-manoalide in the inhibition of bovine PLA₂ and discuss the chemical structure of manoalide responsible for the enzyme inhibition.

The inhibition activities of *d,l*-manoalide, seco-manoalide, and analogues 1a, 1b, 1c, and 2a were estimated by measuring the residual PLA₂ activity towards the micelles of dilauroylphosphatidylcholine mixed with an anionic surface-active agent, sodium cholate, since the hydrolytic activity of bovine PLA₂ is significantly activated by the presence of anionic surface-active agents, unlike snake venom PLA₂s. The results are summarized in Figs. 1, 2, and 3.⁷ As seen from Figs. 2 and 3, an analogue of 1a exhibited inhibition comparable to that by seco-manoalide. An analogue of 2a was found to be an inhibitor as potent as manoalide. It is of interest

Fig 1



that seco-manoalide is a stronger inhibitor than manoalide, and that an analogue of $\underline{1c}$ showed no inhibitory activity (Fig.1). Thus, compounds $\underline{1a}$ and $\underline{2a}$ were concluded to be simple and effective analogues of seco-manoalide and manoalide.

In order to determine the chemical structure of the manoalide molecule responsible for the enzyme inhibition, the inhibitory activities of homologues $\underline{3}$,⁸ $\underline{4}$, and $\underline{5}$ ⁹ were measured by the method described above. Monoaldehyde derivatives $\underline{3}$ and $\underline{4}$ showed no inhibitory activity at all. Dehydroxy derivative $\underline{5}$ showed a much weaker inhibitory activity than analogue $\underline{1a}$. These results indicate that the two aldehyde groups of manoalide are indispensable to inhibit the enzymatic activity of bovine pancreatic PLA₂. The same conclusion was also obtained for cobra⁴ and bee⁵ venom PLA₂s.

The synthesis of analogues $\underline{1a}$, \underline{b} , \underline{c} , $\underline{2a}$, and $\underline{4}$ was performed by our highly effective methods to prepare manoalides as described previously (Fig. 4).¹⁰ Thus, allyl chlorides $\underline{7a}$, \underline{b} , and \underline{c} were prepared from the corresponding halides $\underline{6a}$, \underline{b} , and \underline{c} by a reaction with the anion of dimethylhydrazone of pyruvaldehyde dimethylacetal followed by acid treatment, Peterson olefination, reduction, and chlorination (40% yield of $\underline{7a}$ from $\underline{6a}$). These allylic chlorides were coupled with carbon monoxide and 2-trimethylsilyl-4-tributylstannylfuran catalyzed by palladium(0) in the presence of triphenylphosphine to give the corresponding ketones. The ketones were reduced with lithium aluminum hydride, and the alcohols were protected to afford silylfuran derivatives $\underline{8a}$, \underline{b} , and \underline{c} (89% yield of $\underline{8a}$ from $\underline{7a}$). Photosensitized oxygenations of $\underline{8a}$, \underline{b} , and \underline{c} followed by acid treatment yielded analogues of seco-manoalide $\underline{1a}$, \underline{b} , and \underline{c} , respectively (60% of $\underline{1a}$ from $\underline{8a}$), through compound $\underline{9}$. Compound $\underline{2a}$ was quantitatively obtained by photoirradiation of $\underline{1a}$. Compound $\underline{4}$ was synthesized from farnesyl chloride by a reaction with the anion derived from 3-(ethoxyethoxy) tributylstannylfuran (71% yield), followed by photosensitized oxygenation in

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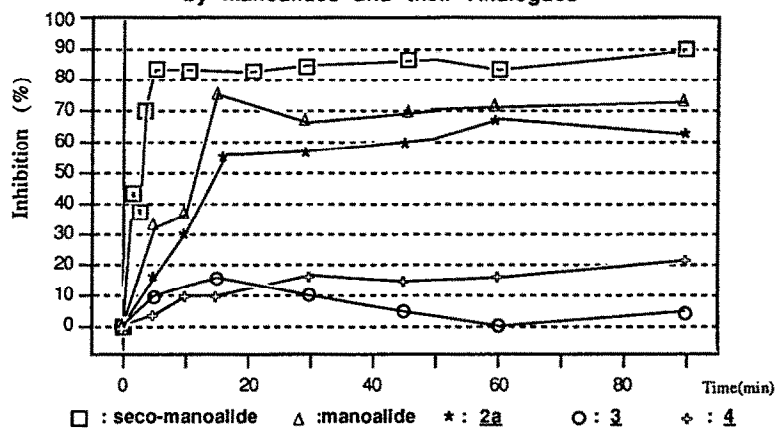


Figure 1 is a line graph showing the inhibition of the reaction of 1a with 1b by various compounds over time. The y-axis represents 'Inhibition (%)' from 0 to 100, and the x-axis represents 'Time(min)' from 0 to 100. The legend indicates five series: seco-manoolalde (squares), 1a (triangles), 1b (asterisks), 1c (diamonds), and 5 (circles).

Time (min)	seco-manoolalde (%)	1a (%)	1b (%)	1c (%)	5 (%)
0	0	0	0	0	0
2	42	35	0	0	0
5	82	38	0	0	0
10	82	32	0	0	28
15	82	58	0	0	35
30	85	75	0	0	72
45	88	85	72	0	78
60	85	82	78	10	75
90	88	85	78	10	78

Fig 4

$$\text{R-X} + \begin{array}{c} \text{MeO} \quad \text{OMe} \\ | \quad | \\ \text{C} \\ | \\ \text{N-NMe}_2 \end{array} \xrightarrow[\text{d), e)]}{\text{a), b), c)]} \text{I} \xrightarrow[\text{h)]}{\text{CO} + \text{f), g)]} \text{II} \xrightarrow{\text{I)}} \text{III} \xrightarrow{\text{j)}} \text{IV} \xrightarrow{\text{k)}} \text{V}$$

$$\text{I} + \begin{array}{c} \text{SiMe}_3 \\ | \\ \text{C} \\ | \\ \text{SnBu}_3 \end{array} \xrightarrow{\text{I)}} \text{VI} \xrightarrow{\text{m), n)}} \text{VII}$$

Chemical reaction scheme showing the synthesis of compound 5 from R-X and 1,1-dimethylethylenediamine (I). The scheme involves several steps: I) reaction with CO and SnBu_3 to form II; II) reaction with I to form III; III) reaction with j) to form IV; IV) reaction with k) to form V; V) reaction with I to form VI; VI) reaction with m, n) to form VII (5).

a) Li, Et₂NH / THF, PhH, HMPA b) H₃O⁺ c) Me₃SiCH₂CO₂Bu^t, LDA / THF, d) DIBAL CH₂Cl₂ e) MsCl, LiCl, collidine / DMF
f) Pd(dba)₂, PPh₃ / THF g) LAH / Et₂O h) CH₃OCH(CH₃)Cl, (iso-Pr)₂NEt / CH₂Cl₂ i) ¹O₂ / MeOH j) 2N-HCl / THF
k) hv / benzene l) BuLi / THF m) ¹O₂ / MeOH n) 2N-HCl / MeOH

the presence of N,N-diisopropylethylamine and then acid treatment¹¹ (45% yield). This alkoxystannane route may open the way for easy supply of **1a** and **2a** because of its simplicity.⁹

Thus, we have obtained a good analogue of manoalide **1a** which can be supplied in sufficient quantity to study the inhibition mechanism of bovine pancreatic PLA₂ by manoalide. We have also revealed that the two aldehyde groups of manoalide are indispensable to the enzyme inhibition. It is of great interest to identify the amino acid residues modified by manoalide analogue, **1a**. In the accompanying paper, it is described that only two of the eleven native lysine residues were modified by **1a** or **2a** under the present conditions.

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References and Notes

- +) A part of this work was reported at the 58th of Annual Spring Meeting of Japan Chemical Society, abstract paper p.1183(1989), and at the 34th Symposium on the Chemistry of Terpenes, Essential Oils, and Aromatics, abstract paper p.213(1990).
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7. Assay conditions are as follows. PLA₂(4.4x10⁻⁶M) was incubated with manoalide or its analogue(4.4x10⁻⁴M) in 50mM Tris-HCl buffer(pH 8.0) in the presence of 8.3%(v/v) dioxane at 40°C for 20 minutes. Residual PLA₂ activity toward the mixed micelles of 1.1 mM dilauroylphosphatidylcholine with 5.4mM sodium cholate in the presence of 10mM CaCl₂ was measured by a pH-stat method at 25°C and pH 7. The titration volumes recorded during the enzymatic hydrolysis of the substrate were corrected by subtracting the volume due to the spontaneous reaction in the absence of the enzyme.
8. Compound **3** was derived from **9** by reduction and acid treatment (NaBH₄ in EtOH, 2N-HCl MeOH). See S.Katsumura, S.Fujiwara, and S.Isoe, *Tetrahedron Lett.*, **28**, 1191(1987).
9. Compound **5** was called manoalog by Dennis.⁴ Our simple and efficient synthesis of **5** will be described elsewhere along with an easier route to **1a** and **2a**.
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